

and it forms hydrogen bonds with highly conserved residues across this interface. Mutating D12 to alanine in BsPFK enhances PEP binding by 100-fold with no change in the extent of PEP inhibition. When D12A was introduced into LbPFK only a 5-fold enhancement in PEP binding was observed. Crystal structures of D12A BsPFK and D12A LbPFK were solved to 2.4 Å resolution. Comparison of D12A and wild-type BsPFK with fructose 6-phosphate bound shows a quaternary shift along the active site interface, breaking the hydrogen bonds involving D12. By contrast, D12A LbPFK exhibits no major change in structure relative to wild type BsPFK. In hopes of further enhancing PEP binding, the following mutations of non-conserved residues in the allosteric site were made to the corresponding residues in either EcPFK or BsPFK, respectively. H59D, E55Y, D187E and S319Y combined showed no enhancement in PEP binding. S211R, D214K and G216S alone and in combination also had no effect on PEP binding. All these mutations suggest that the diminished PEP binding affinity to LbPFK is the consequence of more than just the residues in the allosteric site, likely involving the resistance of this enzyme to undergoing the quaternary shift. Funding came from NIH grant GM33216, NIH CBI training grant, and the Welch Foundation grant A1543.

#### 2289-Pos Board B259

##### Kinetics and Thermodynamics of the Interaction of ANS with Proteins

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1-anilino-naphthalene-8-sulfonate (ANS) is a fluorescent probe widely used in protein folding and conformational transitions studies. The fluorescent features of ANS, a blue shift of the emission maximum and the increase of quantum yield and lifetime, are generally attributed to the binding at hydrophobic sites. Despite the interaction of ANS with proteins has been extensively studied since the early works of Gregorio Weber, few high-resolution structures of proteins complexed with ANS have been resolved. In this work the binding of ANS to BSA was analyzed at equilibrium and pre-equilibrium conditions. The combined analysis of fluorescence, near UV circular dichroism and isothermal titration calorimetric data provided a detailed description of the binding mechanism. Three ANS molecules bound to BSA in 100 mM phosphate pH 7 at 25°C. Pre-equilibrium experiments allowed to determine the affinity and the relative quantum yield at each binding site by fitting a microscopic model to the fluorescence time-course data. This analysis unambiguously indicated that the binding of ANS to BSA occurs at two different and independent binding sites with similar quantum yields and affinities ( $\Delta G^\circ$  @ -35 kJ/mol). The binding of ANS to the first site is thermodynamically favored by similar contributions of the enthalpic ( $\Delta H = -16.3$  kJ/mol) and entropic terms ( $-T\Delta S = -19.4$  kJ/mol), while the binding to the second site is enthalpically driven ( $\Delta H = -36.6$  kJ/mol;  $-T\Delta S = 4.3$  kJ/mol). Complementary information from molecular docking showed 3 ANS molecules bound at hydrophobic cavities in BSA subdomains IIA and IIIA with binding affinities in the order of those found experimentally. The sulfonate group of ANS was oriented towards clusters of polar residues, a common feature in the reported crystal structures of other ANS-protein complexes.

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#### 2290-Pos Board B260

##### Investigation of the Ligand-binding Mechanism of Methionine Sulfoxide Reductase A of *E. coli*

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The rise of free oxygen in the atmosphere over 2.5 billion years ago made it possible for large land-based plants and animals to thrive. But oxygen, and the energy it provides, comes at great cost. Aerobic metabolism generates highly reactive intermediates and by-products in the form of hydroxyl radicals, superoxide anions and hydrogen peroxide. Within the cell, these reactive oxygen species attack biological macromolecules, producing covalent modifications that can affect both function and structure. One amino acid residue in proteins that is particularly sensitive to oxidation is the sulfur-containing side chain of methionine. Fortunately, methionine oxidation can be reversed by the actions of peptide methionine sulfoxide reductase (MsrA), which reduces methionine sulfoxide back to methionine and restores function to damaged proteins. We have used multiple spectroscopic techniques to investigate the mechanism by which MsrA recognizes and binds to a wide range of oxidized substrates in need of repair. Substrates studied include proteins, peptides and the non-steroidal anti-inflammatory drug Sulindac. Competition experiments with the fluorescent reporter ANS suggest the existence of weak, but specific, hydrophobic interactions between MsrA and unstructured and/or hydrophobic ligands.

#### 2291-Pos Board B261

##### Characterization of and Kinetics studies on Lipid Extraction of GM2AP Tryptophan Mutants using Intrinsic Fluorescence and a Dansyl-Based Fluorescence Assay

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GM2AP is an accessory protein that functions as a co-factor in degradation of the GM2 ganglioside to GM3. This non-enzymatic lipid transfer protein solubilizes GM2 from intralysosomal vesicles for reaction with HexA. The precise molecular interactions and method of extraction of the GM2 ganglioside from the lipid membrane are not yet known. GM2AP contains four disulfide bonds and three tryptophan residues (W5, W63, W131) with two of these (W63, W131) located in putative membrane binding loops. In this report, the intrinsic tryptophan fluorescence of a series of single and double TRP mutants (W5A, W5AW63A and W5AW131A) of GM2AP is used to characterize protein in solution and in the presence of lipid vesicles. Additionally, results from quenching experiments are shown, where the fractional accessibility of each tryptophan is determined for both neutral and acidic solutions. The kinetics of lipid transfer of each of the tryptophan mutants were also assayed for their ability to extract and transfer dansyl-labeled lipids from liposomes. Removal of the TRP moieties from the putative membrane binding loops results in slower lipid extraction rates, implying that these residues are important in the membrane binding of GM2AP.

#### 2292-Pos Board B262

##### Binding of Organochloride and Pyrethroid Pesticides To Estrogen Receptors $\alpha$ and $\beta$ : A Fluorescence Polarization Assay

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Several agricultural pesticides and industrial chemicals, such as the organochlorides DDT and methoxychlor, have been shown to cause both endocrine disruption in humans and binding to the estrogen receptor. Estrogen receptor- $\alpha$  (ER $\alpha$ ) and - $\beta$  (ER $\beta$ ) are ligand-activated nuclear signaling proteins with widespread presence in the body. Binding of the hormone estradiol to the ER can affect development with an activation profile that is subtype specific. This investigation measures the ability of pesticides to bind *in vitro* to human ER $\alpha$  and ER $\beta$  by observing their ability to displace a fluorescent estrogen homologue from the receptor. Eight pesticide related compounds were assayed: four in the DDT family and four based on the natural insecticide pyrethrin. The organochlorides tested were: DDT, HPTE, and two DDT metabolites: VF77-land VF72-1. The four synthetic pyrethroids tested were permethrin, deltamethrin, bifenthrin and fenvalerate. While all of the molecules in the DDT family that we tested showed strong binding to ER $\alpha$ , the pyrethroids showed either extremely weak (fenvalerate) or no binding (permethrin, deltamethrin, bifenthrin) to ER $\alpha$  under our experimental conditions. ER $\beta$  exhibited a different binding profile: high affinity binding to the DDT family of molecules AND to permethrin, lower affinity but still strong binding to deltamethrin and fenvalerate, and no binding at all to bifenthrin. These results suggest that permethrin, in addition to the DDT based molecules, could potentially have the ability to disrupt the estrogen hormone pathway through binding to ER $\beta$ . Permethrin's binding to ER $\beta$  is notable, particularly in light of its widespread use in home pet care products such as pet shampoo and flea and tick repellants. The results also suggest that the binding affinity of ER $\beta$  is similar to but less discriminating than that of ER $\alpha$ .

#### 2293-Pos Board B263

##### Characterization of the Ca<sup>2+</sup> Binding Affinity and Coordination Site of the LIN-12/Notch-Repeat

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Notch receptors are transmembrane glycoproteins of a highly conserved signaling pathway that regulate cell growth, differentiation, and death in multicellular organisms. Notch activation requires two successive ligand-induced proteolytic cleavages that enable the intracellular Notch to translocate to the nucleus and regulate gene transcription. Notch proteins exhibit a highly conserved modular architecture, which includes three tandem LIN-12/Notch-Repeats (LNRs) responsible for maintaining the receptor in its resting conformation prior to ligand binding. These highly conserved modules contain a characteristic arrangement of three disulfide bonds and a group of aspartate/asparagine residues that coordinate a Ca<sup>2+</sup> ion, essential for the correct folding of an LNR. Outside of the Notch family of proteins, LNR modules also exist in proteins such as the PAPP-A and the stealth proteins. In our previous work, we had recombinantly expressed, purified, and refolded the first repeat of human Notch1 and used it as a model system to characterize the binding specificity and affinity of different